

Risk Assessment and Environmental Geochemistry of Pb, Cu and Fe in Surface Sediments (Case Study: Hashilan Wetland, Kermanshah, Iran)

Samar Mortazavi^{1*}, Behnaz Attaeian², Shoayb Abdolkarimi³

¹Assistant Professor, Department of Environment, Faculty of Natural Resource and Environmental Sciences, Malayer University, Malayer, Iran

²Assistant Professor, Department of Range and Watershed Management, Faculty of Natural Resource and Environmental Sciences, Malayer University, Malayer, Iran

³M.Sc. Graduated, Department of Environment, Faculty of Natural Resource and Environmental Sciences, Malayer University, Malayer, Iran

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ABSTRACT The total concentration and percentage of heavy metals content in the sediment of Hashilan wetland, Kermanshah province, was assessed in summer 2015. The total concentrations of Cu, Pb, and Fe were around 18.57, 6.95 and 379.24 $\mu\text{g g}^{-1}$, respectively. Successive extractions of the samples showed a decreasing trend as follows: residual>organic matter>iron and manganese oxides>exchangeable parts. The comparison of the total metal concentrations and SQGs and NOAA standards proposed high concentrations of Pb and Fe contents, while RAC suggested that Cu, Pb, and Fe had posed a low-moderate risk, low risk and no risk to the system, respectively. Furthermore, the results showed that the most common sources of Pb and Fe were natural while sources of Cu was anthropogenic. Continuous monitoring of the wetlands, particularly water inputs are recommended.

Key words: Hashilan wetland, Surface sediments, Heavy metals, Fractionation

1 INTRODUCTION

The value of wetlands to human society consists of several physical, chemical and biological functions. The basic components of aquatic systems, including water, soil, plants, animals, and nutrients constitute the uniform system of wetlands (Mansoori, 2007). Being potentially toxic to creatures, nowadays heavy metals are the most important life threatening contaminants in aquatic environments (Pasyna and Nriago, 1998; MdSaifullet *et al.*, 2014), especially through food chains (SabzAlizade

and Naseri, 2011). Sediments, which are usually subject to pollution threats from natural and human activities (Gorkem and Duyuse, 2008), are accounted as the biggest sink for contaminants and play important roles in accumulating some heavy metals in the benthic invertebrates and transferring them to higher levels of food chains (Sayadi *et al.*, 2015; Shirneshan *et al.*, 2013; Yap *et al.*, 2002; Long, 1996; Fichet *et al.*, 1999). Therefore, monitoring sediment could be considered as an appropriate way to assess contamination in

*Corresponding author: Assistant Professor, Department of Environment, Faculty of Natural Resource and Environmental Sciences, Malayer University, Malayer, Iran, Tel: +98 9166652008, E-mail: mortazavi.s@gmail.com

aquatic environments (Gagnou and Fisher, 1997), and often used to identify the contamination source in both spatial and temporal scales to involve long-term geochemical processes (Birch *et al.*, 2001; Shirneshan *et al.*, 2013).

Translocation of heavy metals from sediments to the upper water resources could occur as a result of biological activities and physiochemical parameters (Hughes and Chester, 1967; Helling *et al.*, 1990; Chen and Gupta, 1990; Karbassi, 1998; Karbassi and Volvi, 2010; Karbassi *et al.*, 2010; Shirneshan *et al.*, 2013). Since several factors are involved in heavy metal contamination, the assessment of the total concentration of heavy metal is not sufficient to get all the required information in assessing the environmental impact of the contaminated sediments (Obbard and Cuong, 2006). As a result, chemical separation studies have become necessary tools to assess the sources of heavy metals (Bagheri *et al.*, 2011; MdSaiful *et al.*, 2014), and studies in this regard have become crucial to proper management of aquatic environment (Hakanson, 1980; Salamons and Forstner, 1984; Calmano *et al.*, 1993; Dickinson *et al.*, 1996; Sharma *et al.*, 1999; Saeedi *et al.*, 2004; Caeiro *et al.*, 2005; Saeedi *et al.*, 2012). Such studies have been extended to various parts of the world, including Iran, that show the extent of heavy metal contaminations and their sources (Duyusen and Gorkem, 2008; Bagheri *et al.*, 2013; Shirneshan *et al.*, 2013; Silinet *et al.*, 2013).

It seems that agricultural practices such as irrigation, the application of fertilizers and pesticides considerably affect the distribution of heavy metals (Silin *et al.*, 2013). Agricultural practices in Hashilan region is common, accelerating heavy metal inputs to the system (Mohammadi *et al.*, 2013). High heavy metals concentration in Hashilan provides a constant

ecological and health risk to the wetland. In order to determine the contributions of natural and anthropogenic sources in transferring heavy metals in the Hashilan wetland, the concentrations of heavy metals in surface sediments were fractionated and evaluated based on the international sediment quality standards of ISQG (Interim Sediment Quality Guidelines -Canada) and NOAA (National Oceanic and Atmospheric Administration).

2 MATERIALS AND METHODS

2.1 Site Description

Hashilan Wetland (46° 15' 54" E and 34° 34' - 34° 35' N) is situated 36 km northwest of Kermanshah Province, west of Iran (Mohammadzade *et al.*, 2013). Sediment samples were randomly collected (in three replications) from 0-10 cm layer at five stations (north, south, east, west and center) in 2015 (Figure 1). The samples were kept cool and transferred to the Environmental Science Lab, Malayer University, in the polyethylene bags for further analysis. The location of the sediment samples was recorded using a GPS (Table 1).

2.2 Preparation of sediment samples

The samples were first oven-dried at 80-95°C for at least 16 hours to reach a constant weight, then crushed in a stone mortar and sieved through a 63µm sieve. Finally, the powdered samples were kept in polyethylene containers at in a refrigerator.

Table 1 Geographical coordinations and descriptions of Hashilan sampling stations

Stations	Geographical coordinates	Sea level (m)	Landuse description
North	34° 35' 1.31" N 46° 53' 26.32" E	1310	Water inlet lagoon
East	34° 34' 32.25" N 46° 53' 28.28" E	1310	Pear tree
Center	34° 34' 33.66" N 46° 52' 56.16" E	1309	Big kool
South	34° 34' 54.39" N 46° 52' 40.12" E	1309	Willow trees
West	34° 34' 39.14" N 46° 52' 19.37" E	1310	Rabkholudarijh

**Figure 1** General location of the study area and spatial pattern of sampling stations

2.3 Analysis of total concentration of Cu, Pb and Fe in sediment samples

First, about 1 g of dried sub-samples were digested in a hot block digester by a 4:1 v/v ratio of nitric acid (69%) and perchloric acid (60%) for 1 and 3h at the temperatures of 40 and 140 °C, respectively. The digested samples were then diluted with double-distilled water to 25 ml and filtered with Whatman filter paper 0.45µm and kept in polyethylene containers in the fridge (Yap et al., 2002; Li et al., 2001).

2.4 Analysis of Cu, Pb and Fe concentrations in various geochemical parts of sediments

Geochemical fractionation of Cu, Pb and Fe in the sediments was done by the use of successive extraction procedure as follows (Yap et al., 2002; Li et al., 2001): 10 g sub-samples was shaken on a shaker with 50 ml of 1M ammonium acetate and pH 7 at room temperature to obtain the exchangeable form. The form bound to oxides and hydroxides of iron and manganese, remaining material from

the previous step was acidized with 50 ml hydroxyl ammonium chloride to pH 2, then placed on the shaker in room temperature for three hours with hydrochloric acid. To investigate the form bounded to organic matter, the remaining of previous stage was first oxidized with 30% H₂O₂ in a water bath at 90-95°C, then cooled down and placed on the shaker for three hours with 0.1 M ammonium acetate pH 2 in room temperature. Finally, to obtain a residual form, the remaining of the third stage was put in a hot block digester with a 4:1 compound of nitric acid (69%) and perchloric acid (60%) at the temperatures of 40 and 140°C for 1 and 3 hours, respectively. The final solution was filtered through Whatman filter paper 41 and diluted with 70 ml distilled water. Concentration of desired elements in geochemical fractionation was measured by the means of Analytic Jena ContraAA 700 atomic absorption spectrometry.

The data were reported in $\mu\text{g g}^{-1}$ of sample dry weight. The instrumental Limit of Detection (LOD) of Fe, Cu and Pb were 0.02, 0.03 and 0.001 $\mu\text{g g}^{-1}$ dry, respectively. Recovery of the AAS method for heavy metals was measured 80- 95%.

The statistical software package version 9.1 (SAS Institute Inc, NC, USA) was used for all statistical analyses. The assumptions of normality and homogeneity of variances were investigated using Shapiro-Wilk and Levene tests, respectively. Then, any significant differences between the average concentrations of heavy metals in different stations were assessed using one-way analysis of variance (ANOVA) procedure and Duncan multiple range test. In addition, any correlation between Cu, Pb, and Fe concentrations was explored using a Pearson correlation test.

2.5 Enrichment factor (EF)

Enrichment factor is computed to estimate the sediment chemistry in relation to anthropogenic and natural contamination sources using the equation (1) (Zonta *et al.*, 1994; Loska *et al.*, 2003; Suthar *et al.*, 2009; Shirneshan *et al.*, 2013).

$$EF (\%) = (C - C_{\text{Min}})/(C_{\text{Max}} - C_{\text{Min}}) \times 100 \quad (1)$$

C: Mean metal concentration in sediments (mg kg^{-1})

C_{Max} and C_{Min} : Maximum and minimum estimated concentrations

2.6 Environmental risk assessment code

Risk assessment code (RAC), first introduced by Perin (1985), and was applied to assess the mobility and bioavailability of heavy metals in surface sediments (Wang *et al.*, 2010).

3 RESULTS

3.1 Heavy metal concentration and percentage in geochemical forms

The mobility, access, and potential toxicity of heavy metals in the sediment were assessed using the concentration and percentage of metals, including Cu, Pb, and Fe (Figure 2). Natural and anthropological sources of Pb, Cu and Fe in Hashian wetland sediment are shown in Figure 3.

3.2 Comparison of metal concentrations and international standards

The heavy metal contents were compared with the global standard values suggested by SOGs (Standard Operating Guidelines) (TEC and PEC indices) and NOAA (ERL and ERM indices) (Table 5). The calculated amounts related to EF and the summary of comparisons are presented in Table 2 and Table 3, respectively.

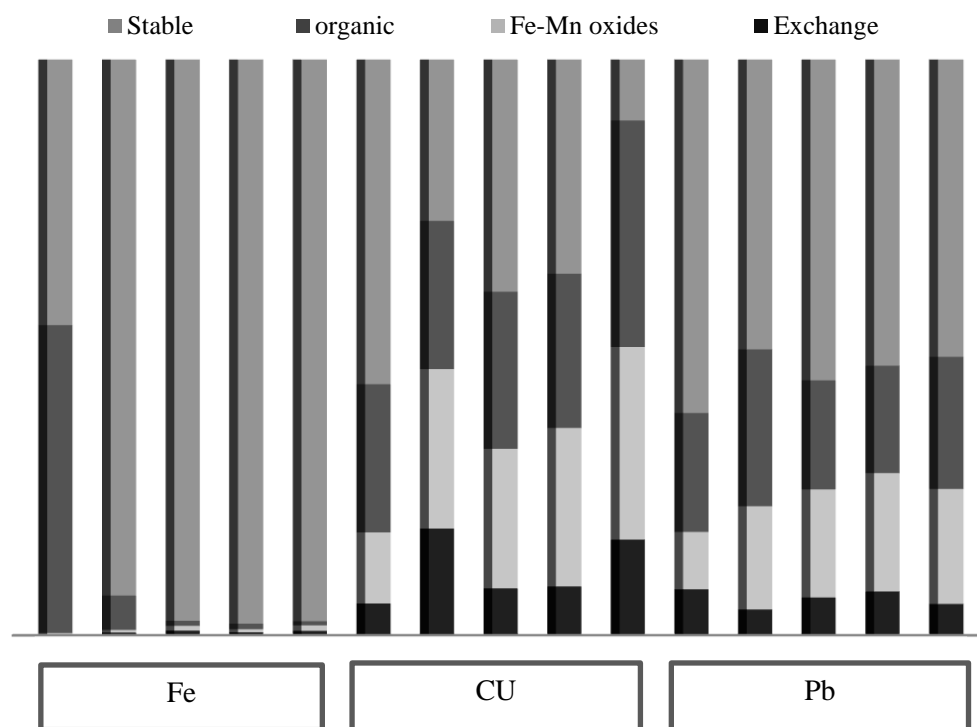


Figure 2 Distribution of Cu, Pb, and Fe percentage in geochemical forms of sediment

Table 2 A summary of enrichment index of different heavy metals in surface sediment

Stations	Replication	EF (Cu)	EF (Pb)	EF (Fe)
N	3	60.21	54.24	73.51
W	3	25.52	19.13	28.71
E	3	30.21	21.71	6.67
S	3	18.67	17.57	11.11
P	3	4.29	9.71	10.05

Table 3. Comparison of heavy metal concentrations with environmental risk assessment code

RAC	Index	Cu	Pb	Fe
No risk	< 1			0.21-0.82
Low risk	1-10	5.55-8.55	4.51-8.04	
Medium risk	11-30	16.78-18.73		
High risk	31-50			
Very high risk	>75			

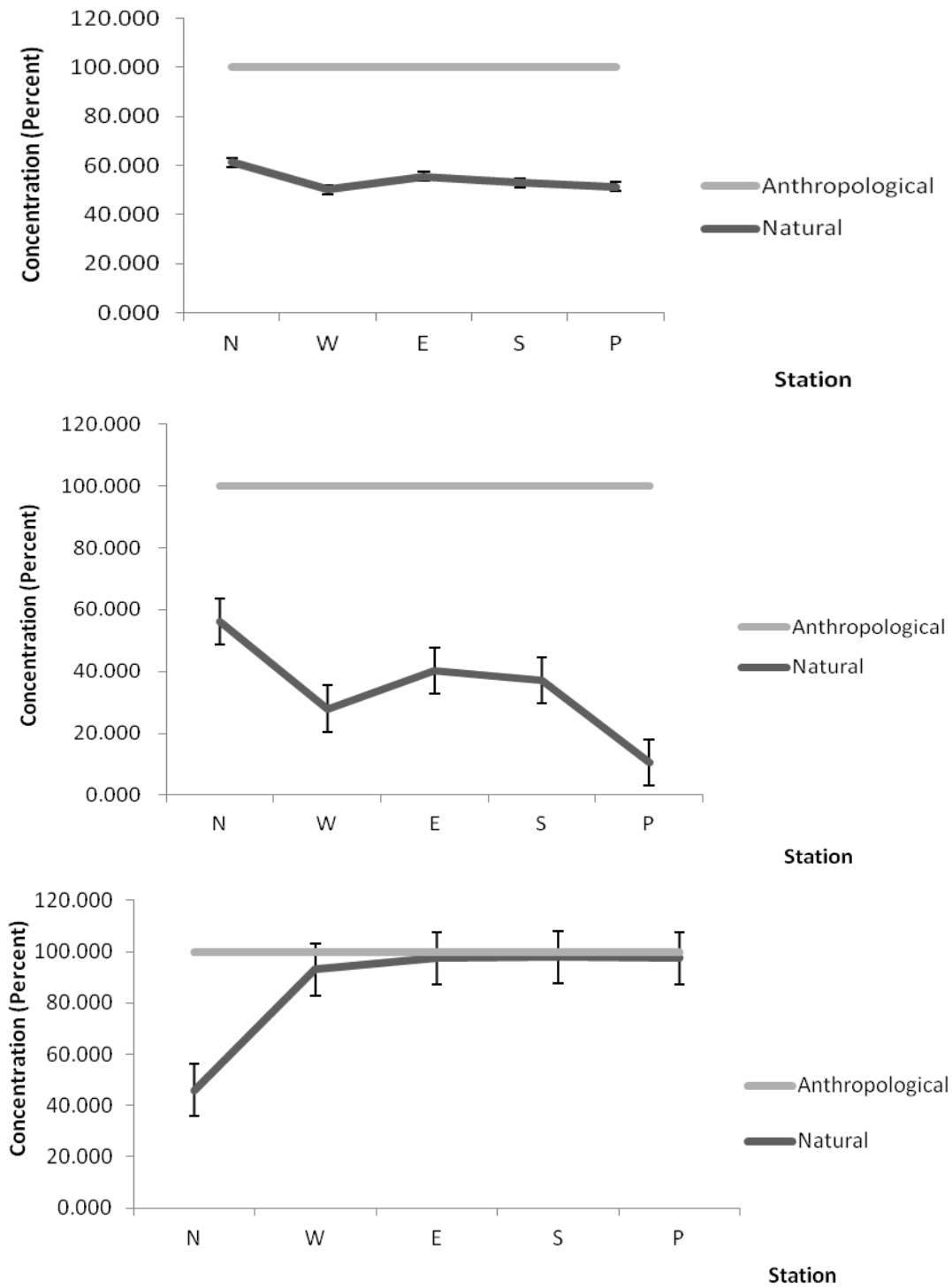


Figure 3 Natural and Anthropological sources of Pb (up), Cu (middle) and Fe (down) in Hashian wetland sediment

4 DISCUSSION

4.1 Concentration and metal percentage in geochemical parts

Considering the average concentrations and percentages of Cu, Pb, and Fe in geochemical forms of the surface sediment and their enrichment index, the maximum ($1609 \mu\text{g g}^{-1}$) and minimum ($1.11 \mu\text{g g}^{-1}$) Cu concentrations at the sampling stations were attributed to the residual part. However, the highest ($5.97 \mu\text{g g}^{-1}$) and lowest ($0.29 \mu\text{g g}^{-1}$) Pb concentrations were related to the residual and exchangeable parts, respectively. The highest ($450.47 \mu\text{g g}^{-1}$) and lowest ($1.23 \mu\text{g g}^{-1}$) Fe concentrations consisted of organic matter and exchangeable part, respectively. The results of successive extractions indicated a decreasing trend of different geochemical parts in the following order: residual > organic matter > iron-manganese oxides > exchangeable. Further, the results also showed a natural source of Cu, Pb, and Fe metals in the sediment samples in different geochemical parts. The main source of Cu included anthropogenic activities (Figs. 3). In order to determine any significant differences in Cu, Fe and Pb concentrations among the stations, analysis of variance and Duncan test were applied. Results of Levene test for their concentrations indicated homogeneity of variances across groups ($p > 0.05$). The results of Duncan revealed that there were significant differences between concentration of Cu in station 1 and 2 and concentration of Fe in station 1 with other stations. No significant differences were observed in Pb concentrations among the sites.

4.2 Enrichment factor

The enrichment factor (EF) values for Cu, Pb and Fe ranged from 4.29-60.21, 9.71-54.24, and 6.67-73.51, respectively. The maximum EFs were observed in the north station of Hashilan wetland, while the minimum EF of Cu and Pb were evident in the central station and for Fe in

the eastern station. The low index in the central station seemed to be due to the existence of reed colonies with high refining ability. The land uses that more severely caused contamination included rangelands, mountains, forests, recreational area of Sabzali Sarab, environment monitoring station, roads, and water transmission channel.

4.3 Comparison of sediment data and other studies

The average concentrations of Cu, Pb and Fe in the surface sediments of Hashilan wetland have been compared with the other studies around the world as follows:

4.4 Comparison of sediment data and international standards

The comparison of the heavy metal contents in this study with the international standards showed no toxicity risk of the Cu as its content was low in all the stations of Hashilan wetland (Table 5). The results of Pb and Fe were different. The average Pb content was low except for ERL (Effect Range Low) standard. It seemed that Fe content was more than the other metals in the wetland and its level was higher than PEL (Probable Effects Levels), ERM (Effects Range Median), PEC (Probable Effects Concentration) and ERL international standards.

Table 4. Comparison of heavy metal concentrations between the sediment samples of the study area and around the world

Local	Cu ($\mu\text{g g}^{-1}$)	Pb($\mu\text{g g}^{-1}$)	Fe($\mu\text{g g}^{-1}$)	Reference
Wetland Anzali, Iran	57.13	27.61	45424	Jamshidy Zanjani and Saedi, 2013
Mangrove forest, Bushehr, Iran	46.04	94.8	30736.72	Davari <i>et al.</i> , 2012
Vigo Ria (NW Iberian Peninsula)	18-82	24-112	18-34	Maria <i>et al.</i> , 2008
Taranto Gulf (Ionian Sea, Southern Italy)	47	59	26313	Alessandro <i>et al.</i> , 2006
	105	219	3.9	Acevedo-Figueroa <i>et al.</i> , 2006
	22	7.6	4.9	
Punta Mala, Bay Panama	56.3	76.2	9825	Defew <i>et al.</i> , 2005
wetlands of Sydney, Australia	9.6	27.61	9422	Connor and Thomas, 2003
Queensland, Australia	3.08	6.25	13583.96	Preda and Cox, 2002
Morgan County, America	2.1	2.5	556	Mays and Edwards, 2001
Goreckie Lake, Wielkopolski National Park, western Poland	12	47.5	9150	Zerbe <i>et al.</i> , 1999
Milos bay, Greece	51	151	34000	Karageorgis <i>et al.</i> , 1998
Bothnian Bay	52	79	52	Leivuori, 1998
Bothnian Sea	36	42	36	
Gulf of Finland	43	50	43	
Caribbean coast Costa Rica	8.4	34.5	8.4	Guzman and Jimenez, 1992
UK estuaries (range)	7-2398	20-2753	7-2398	Bryan and Langston, 1992
Gulf of Venice, Italy	17	38	17	Donazzolo <i>et al.</i> , 1981
Hashilan Wetland	18.57	6.95	18.57	Current study

Table 5. Comparison of heavy metal concentrations with international standards

Refs	Fe	Pb	Cu	Standards
TEC	31.6	35.8	-	Hongyi, 2009; Mooraki et al, 2009
PEC	149	459	459	Galindo-Riano et al, 2009
PEL	108	112	271	CCME, 1999; Mooraki et al., 2009; Galindo-Riano et al., 2009
ERM	270	218	410	Long, 1995; Mooraki et al., 2009; Galindo-Riano et al., 2009
ERL	34	7.46	150	Long, 1995; Mooraki <i>et al.</i> , 2009; Galindo-Riano <i>et al.</i> , 2009
-	18.57	6.95	379.24	Current Study

5 CONCLUSION

The total contents of heavy metals Cu, Pb, and Fe were separately assessed in different geochemical parts of the sediment samples in the Hashilan wetland. Cu concentration decreased in the following order: north>east>west>south>center; the maximum content of Cu in the north station was mainly due to the different types of land uses in the station, while its minimum content was observed in the central station where human

interference was relatively low. The maximum and minimum Pb content was also observed in the north and central stations, respectively, as follows: north>east>west>south>center. However, the maximum and minimum Fe contents were measured in the north and east stations, respectively, with the order of north>west>south>center>east. Results indicated the natural origins for Pb and Fe, while anthropogenic origin for Cu.

The comparison of exchangeable part with RAC index showed no risk, low risk, and low-moderate risk for Fe, Pb, and Cu, respectively.

5 REFERENCES

- Abu-Kukati, Y. Heavy Metals Distribution and Speciation in Sediments from Ziqlab Dam – Jordan. *Jeafaş Mühendisliđi*. 2001; 25(1): 33-40.
- Acevedo-Figueroa, D., Jimenez, B.D. and Rodriguez-Sierra, C.J. Trace metals in sediments of two estuarine lagoons from Puerto Rico. *Environmental Pollution*. 2006; 14: 336-342.
- Alessandro, B., Giovanni, B., Nicola, C., Angelo, D. A., Antonella, D. L. and Antonella, M. Heavy metals in marine sediments of Taranto Gulf (Ionian Sea, Southern Italy). *Marine Chemistry*. 2006; 99: 227–235.
- Alvarez-Iglesias, P., Rubio, B. and Perez-Arlucea, M. Reliability of subtidal sediments as “geochemical recorders” of pollution input: San Simo’n Bay (Ria de Vigo, NW Spain). *Estuarine, Coastal and Shelf Science*. 2006; 70: 507-521.
- Baghari, H., Sharmd, T., Kherabadi, V., Darvishbastami, K. and Baghari, Z. Tracing and Assessment of Heavy Metals in Gorganrud Sediments. *Oceanography*. 2011; 5(2): 35-39.
- Baghari, Z., Riahi Bakhtiari, A. and Baghari, H. Studies on distribution and fractionation of heavy metals in Surface Sediments of Bandar Abas coastal. *Oceanography*. 2012; 14(4): 27-33.
- Birch, G.F., Taylor, S.E. and Matthai, C. Small-scale spatial and temporal variance in the concentration of heavy metals in aquatic sediments: a review and some new concepts. *Environmental Pollution*. 2001; 113: 357–372.
- Caeiro, S. Assessing heavy metal contamination in Sado Estuary sediment: An index analysis approach. *Ecological Indicators*. 2005; 5: 151–169.
- Calmano, W., Hong, J. and Förstner, U. Binding and mobilization of heavy metals in contaminated sediments affected by pH and redox potential. *Water Science*. 1993; 28: 53–58.
- CCME (Canadian Council of Ministers of the Environment). Canadian environmental quality guidelines. Canadian Council of Ministers of the Environment Winnipeg. 1999.
- Chakarabarty, R., Zaman, S., Mukhopadhyay, N., Banergee, K. and Matra, A. Seasonal variation of Zn, Cu and Pb in the esuarine stretch of West Bengal. *Indian Journal of Marine Sciences*. 2009; 38(1): 104-109.
- Chester, R. and Hughes, M.A. Chemical Technique for the Separation of Ferromanganese Minerals, Carbonate Minerals and Adsorbed Trace Elements from Pelagic Sediments. *Journal of Chemical Geology*. 1967; 2: 242-262.
- Christophoridis, C., Dedepsidis, D. and Fytianos, K. Occurrence and distribution of selected heavy metals in the surfce sediments of Thermaikos Gulf, N. Greece. Assessment using pollution indicators. *J. Hazardous Materials*. 2009; 168: 1082-1091.
- Connor, S.E. and Thomas, I. Sediments as archives of industrialization: Evidence of atmospheric pollution in coastal wetlands of southern Sydney, Australia.

- Water, Air, and Soil Pollution. 2003; 149: 189–210.
- Cuong, D.T. and Obbard, J.P. Metal speciation in coastal marine sediments from Singapore using a modified BCR-sequential extraction procedure. *Appl Geochem.* 2006; 21(8): 1335 -1346.
- Davari, A., khorasani, N. A. and Dana kar, A. Comparison of Heavy Metal Concentration in Bidekhun, Basatin and Melgonze Mangrove Forests. *Iranian Journal of Applied Ecology.* 2012; 2(1): 15-25.
- Defew, L. H., Mair, J. M. and Guzman, H. M. An assessment of metal contamination in mangrove sediments and leaves from Punta Mala Bay, Pacific Panama. *Marine Pollution Bulletin.* 2005; 50: 547–552.
- Dickinson, W.W., Dunbar, G.B. and McLeod, H. Heavy metal history from cores in Wellington Harbour, NewZealand. *Environmental Geology.* 1996; 27: 59–69.
- Dollar, N. L., Souch, C. J., Filippelli, G. M. and Mastalerz, M. Chemical fractionation of metals in wetland sediments: Indiana Dunes National Lake shore. *Environmental Science and Technology.* 2001; 35(18): 3608–3615.
- Donazzolo, R., Merlin, O.H., MenegazzoVitturi, L., Orio, A.A., Pavoni, B., Perin, G. and Rabitti, S. Heavy metal contamination in surface sediments from the Gulf of Venice, Italy. *Marine Pollution Bulletin.* 1981; 12: 417–425.
- Duyusen G. E., GorkemA. Heavy metals partitioning in the sediments of Izmir Inner Bay. *Journal of environmtal Sciences.* 2008; 35(4): 413-418.
- Fernandes, H. M. Heavy metal distribution in sediments and ecological risk assessment: the role of diagenetic processes in reducing metal toxicity in bottom sediments. *Environmental Pollution.* 1997; 97(3): 317–325.
- Fichet, D., Bowher, G., Radenac, G. and Miramand, P. Concentration and mobilisation of Cd, Cu, Pb and Zn by meio fauna populations living in harbour sediment: their role in the heavy metal flux from sediment to food web. *The Science of the Total Environment.* 1999; 243(15): 263-272.
- Galindo-Riano, M.D., Vicente-Martorell, J.J., Garcia-Vargas, M. and Granado-Castro, M.D. Bioavalability of heavy metals monitoring water, sediments and fish species from a polluted estuary. *Journal Hazardous Materials.* 2009; 162: 823-836.
- Gognou, C. and Fisher, N.S. The bioavailability of sediment bound Cd, Co, and Ag to the mussel *Mytilusedulis*. *Canadian Journal Fish Aquatic science.* 1997; 54: 147-156.
- Gupta, S.K. and Chen, K.Y. Partitioning of Trace Metals in Selective Chemical Fraction of Near shore Sediments. *Journal of Environmental Letters.* 1990; 10: 129-158.
- Guzman, H. M. and Jimenez, C.E. Contamination of coral reefs by heavy metals along the Caribbean coast of Central America (Costa Rica and Panama). *Marine Pollution Bulletin.* 1992; 24: 554–561.
- Hakanson, L. Ecological risk index for aquatic pollution control, A sedimentological approach. *Water Research.* 1980; 14: 975–1001.

- Helling, D. Sediments and Environmental Geochemistry. Springer Verlag, New York.1990.
- Hashemi, S. J., RiahiBakhtiari, A. and Lak, R. Source Identification and Distribution of Lead, Copper, Zinc, Nickel, Chromium and Vanadium in Surface Sediments of Caspian Sea. Journal Mazandaran University Medical Science. 2013; 23(1): 36-50.
- Hongyi, N., Wenjing, D., Qunhe, W. and Xingeng, C. Potential toxic risk of heavy metals from sediment of the Pearl River in South China. Journal of Environmental Sciences. 2009; 21(8): 1053–1058.
- JamshidiZanjani, A.andSaedi, M. Metals Pollution Assessment of Surface Sediment in Anzali Wetland and Their Quality Zonation. Journal of Environmental Studies. 2014; 68(4): 157-170.
- Jones, J. M. Chemical fractionation of copper, lead and zinc in Ombrotrophic Peat. Environmental Pollution.1987; 48(2): 131–144.
- Karageorgis, A., Anagnostou, Ch., Sioulas, A., Chronis, G. and Papathanassiou,E. Sediment geochemistry and mineralogy in Milos Bay, SW Kyklades, Aegean Sea, Greece. Journal of Marine Systems. 1998; 16: 269– 281.
- Karbassi, A., NabiBidhendi, G.R., Saedi, M. and Rastegari, A. Metals removal during estuarine mixing of Arvand River water with the Persian Gulf water, Central European Journal of Geosciences. 2010; 2(4): 531-536.
- Karbassi, A.R. Geochemistry of Ni, Zn, Cu, Pb, Co, Cd, V, Mn, Fe, Al and Ca in Sediments of North Western Part of the Persian Gulf. International Journal Gidrometeoizdat. 1998; 360.
- Karbasi, A. A. and Voloi, S. Determination of heavy metal pollution (Hg, Pb, Cd, Ni, Fe and Zn) in the Bamdezh wetland sediments using Muller geochemical index. 2010; 2(54): 1-10
- Leivuori, M. Heavy metal contamination in surface sediments in theGulf of Finland and comparison with the Gulf of Bothnia. Chemosphere. 1998;36: 43-59.
- Li, X., Shen, Z., Wai, O. W. H. and Li, Y. Chemical forms of Pb, Zn and Cu in the sediment profiles of the Pearl River Estuary. Marine Pollution Bulletin. 2001; 42(3): 215-223.
- Long, E.R., Robertson, A., Wolfe, D.A., Hameedi, J. and Sloane, G.M. Estimates of the spatial extent of sediment toxicity in major US estuaries. Environmental Science & Technology. 1996; 30(12): 3585–3592.
- Loska, K. and Wiechula, D. Application of principal component analysis for the estimation of source heavy metal contamination in surface swdiments from Rybnik Reservoir. J. Chemosphere. 2003; 51: 723-733.
- Maria Jesus Belzunce, S., Ricardo, P., M. Jeff, W., Jeff, B. and Juan, S.E. Metal speciation in surface sediments of the Vigo Ria (NW Iberian Peninsula). Scientia Marina. 2008; 72(1): 119-126.
- Mays, P.A. and Edwards, G.S. Comparison of heavy metal accumulation in a natural wetland and constructed wetlands receiving acid mine drainage. Ecological Engineering. 2001; 16: 487–500.

- MdSaiful, I., MdKawser, A., Mohammad, R., MdHabibullah, A. M. and Muhammad, K.I. Heavy metal pollution in surface water and sediment: A preliminary assessment of an urban river in a developing country. *Ecological Indicators*. 2014; 48: 282–291.
- Mooraki, N., Esmaeli Sari, A., Soltani, M. and Valinassab, T. Spatial distribution and assemblage structure of macrobenthos in a tidal creek in relation to industrial activities. *International Journal. Environmental Science and Technology*. 2009; 6: 651-662.
- Mohamadi, M., Karbasi, A. and Sahebi, M. Hashilan wetland environmental management SWOT method. *Journal of Environmental advocacy*. 2013; 57: 15-27.
- Nriagu, J.O. and Pacyna, J.M. Quantitative assessment of worldwide contamination of air, water and soils by trace metals. *Nature* 333. 1988; 134-139.
- KabataPendias, A. Trace Elements in Soil and Plants, 4th ed. 2011. Boca Raton London, New York FL 33487-2742, CRC Press.
- Perin, G., Craboledda, L., Lucchese, M., Cirillo, R., Dotta, L., Zanetta, M. L. and Oro, A. A. Heavy metal speciation in the sediments of Northern Adriatic Sea- a new approach for environmental toxicity determination, In: *Heavy Metal in the Environment*. (ed.) T. D. Lekkas. 1985; 2: 454-456.
- Preda, M. and Cox, M.E. Trace metal occurrence and distribution in sediments and mangroves, Pumicestone region, southeast Queensland, Australia. *Environment International*. 2002; 28: 433–449.
- Shirneshan, G., Riahibakhtiari, A., Sayfabadi, S. J. and Mortazavi, S. Environmental Geochemistry of Cu, Zn and Pb in Sediment from QeshmIsland_Persian Gulf, Iran: a Comparison between the Northern and Southern Coast and Ecological Risk. *Geochemistry International*. 2013; 5(8): 670–676.
- Saeedi, M. Sorbed metals fractionation and risk assessment of release in river sediment and particulate matter. *Environmental Monitoring and Assessment*. 2012; 185: 1737–1754.
- Saeedi, M., Hosseinzadeh, M. and Rajabzadeh, M. Competitive heavy metals adsorption on natural bedsediments of Jajrood River, Iran. *Environmental Earth Sciences*. 2011; 62(3): 519–527.
- Saeedi, M., Daneshvar, S. and Karbassi, A. R. Role of riverine sediment and particulate matter on adsorption of heavy metals. *International journal of Environmental Science and Technology*. 2004; 1(2): 135–140.
- Salomons, W. and Forstner, U. *Metals in the Hydrocycle*. Springer-Verlag, Berlin Heidelberg. 1984; 93.
- Sayadi, M. H., Rezaei, M. R., RashkiGhaleno, O., Afsari., K. and Poor Mollaei., N. Natural and Concentration Factor Distribution of Heavy Metals in Sediments of ChahNimeh Reservoirs of Sistan, *Ecopersia* 2015; 3 (2), 1003-1012
- Silin, Y., Dequn, Z., Huayong, Y., Rong, W. and Bo, P. Distribution and speciation of metals (Cu, Zn, Cd, and Pb) in agricultural and non-agricultural soils near a stream up river from the Pearl River, China. *Environmental Pollution*. 2013; 177: 64-70.

- Kuttickat Paul, S. Geochemistry of Bottom Sediments from a River-Estuary-Shelf Mixing Zone on the Tropical Southwest Coast of India. *Bulletin of Geological Survey of Japan*. 2001; 52 (8): 371-382.
- Sharma, V. K., Rhudy, K. B., Koenig, R., Baggett, A.T., Hollyfield, S. and Vazquez, F.G. Metals in sediments of Texas estuaries, USA. *Journal of Environmental Science and Health*. 1999; 34: 2061–2073.
- Suthar, S., Nema, A.K., Chabukdhara, M. and Gupta, S.K. Assessment of metals in water and sediments of Hindon River, India: Impact of industrial and urban discharges. *Journal of Hazardous Materials*. 2009; 171: 1088-1095.
- Wang, S., Jia, Y., Wang, S., Wang, X., Wang, H., Zhao, Z. and Liu, B. Fractionation of heavy metals in shallow marine sediments From Jinzhou Bay, China. *Journal of Environmental Sciences*. 2010; 22(1): 23-31.
- Yap, C K, Ismail A, Tan S. G. and Omar, H. Correlations between speciation of Cd, Cu, Pb and Zn in sediment and their concentrations in total soft tissue of green-lipped mussel *Perna perna* from the west coast of Peninsular Malaysia. *Environment International*. 2002; 28(1-2): 117–126.
- Yap, C. K., Ismail, A., Tan, S. G. and Omar, H. Concentrations of Cu and Pb in the offshore and intertidal sediments of the west coast of Peninsular Malaysia. *Environment International*. 2002; 28(8): 467– 479.
- Zerbe, J., Sobczyński, T., Elbanowska, H. and Siepak, J. Speciation of Heavy Metals in Bottom Sediments of Lakes. *Journal of Environmental Studies*. 1999; 8: 331-339.
- Zonta, R., Zaggia, L. and Argres, E. Heavy metal and grain size distribution in estuarine shallow water sediments of the Cona Marsh (Venice Lagoon, Italy). *J. Science Direct Total Environmental*. 1994; 15: 19-28.

ارزیابی خطر و تفکیک ژئوشیمیایی فلزات سنگین (مس، سرب و آهن) در رسوبات سطحی (مطالعه موردی): تالاب هشیلان، کرمانشاه، ایران)

ثمر مرتضوی^۱، بهناز عطائیان^۲ و شعیب عبدالکریمی^۳

- ۱- استادیار گروه محیط زیست، دانشکده منابع طبیعی و محیط زیست، دانشگاه ملایر.
- ۲- استادیار گروه مرتع و آبخیزداری، دانشکده منابع طبیعی و محیط زیست، دانشگاه ملایر.
- ۳- دانشجوی کارشناسی ارشد رشته محیط زیست، دانشکده منابع طبیعی و محیط زیست، دانشگاه ملایر.

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چکیده بررسی میزان فلزات سنگین تالاب هشیلان (استان کرمانشاه) در تابستان ۱۳۹۴ با نمونه برداری از رسوبات سطحی آن در ۵ ایستگاه با ۳ تکرار انجام گرفت. غلظت کلسیم و همچنین غلظت فلزات سنگین در بخش های ژئوشیمیایی رسوبات باروش استخراجی پیماندازه گیری گردید. میانگین غلظت کل مس، سرب و آهن به ترتیب ۱۸/۵۷، ۶/۹۵ و ۳۷۹/۲۴ میکروگرم بر گرم تعیین گردید. نتایج استخراج پی در پی در کلیه ایستگاه ها بیانگر روند نزولی بخش های پایدار، آلی، اکسیدهای آهن- منگنز و تبادل می باشد. مقایسه غلظت کلی فلزات با استانداردهای جهانی SQGs و NOAA، بالا بودن غلظت فلزات سرب و آهن را نشان می دهد. در همین راستا نتایج حاصل از کد ارزیابی ریسک، فلز مس را در محدوده ریسک کم تا متوسط، فلز سرب را در محدوده ریسک کم و فلز آهن را در وضعیت بدون ریسک نشان داد. نتایج به دست آمده از سهم فلزات در بخش های مختلف ژئوشیمیایی رسوبات، بیانگر غالب بودن منشأ طبیعی برای دو فلز سرب و آهن، و منشأ انسانی برای فلز مس می باشد. پایش مداوم محیط زیست تالاب بخصوص در بخش آب های ورودی پیشنهاد می گردد

کلمات کلیدی: تالاب هشیلان، رسوبات سطحی، فلزات سنگین، منشأیابی.